

APPLICATION  
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TITLE: DESIGNED PROTEIN PORES AS COMPONENTS FOR  
BIOSENSORS

APPLICANT: HAGAN BAYLEY, ORIT BRAHA, JOHN KASIANOWICZ  
AND ERIC GOUAUX

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DESIGNED PROTEIN PORES AS COMPONENTS FOR BIOSENSORS

Statement as to Federally Sponsored Research

5        This application claims priority from provisional  
application 60/053,737, filed July 25, 1997, which is  
incorporated herein by reference in full.

      This invention was made with U.S. Government support  
under the Office of Naval Research grant No. N00014-93-1-  
10    0962. The government has certain rights in the invention.

Background of the Invention

      The field of the invention is metal detection.

      Biosensors are analytical devices that convert the  
concentration of an analyte into a detectable signal by  
15    means of a biologically-derived sensing element. Well-known  
biosensors include commercial devices for sensing glucose.  
In addition, true biosensors, biomimetic devices, and  
devices that use living cells have recently been developed.  
For example, to detect divalent metal cations, true  
20    biosensors have been made using the enzyme carbonic  
anhydrase (Thompson et al., 1993, Anal. Chem. 65:730-734),  
the metal binding site of which has been altered (Ippolito  
et al., 1995, Proc. Natl. Acad. Sci. USA 92:5017-5020). To  
monitor HIV antibody levels, the enzyme alkaline phosphatase  
25    into which an HIV epitope has been inserted has been  
utilized (Brennan et al., 1995, Proc. Natl. Acad. Sci. USA  
92:5783-5787).

Summary of the Invention

      The invention features a mutant staphylococcal alpha  
30    hemolysin ( $\alpha$ HL) polypeptide containing a heterologous metal-  
binding amino acid. The polypeptide assembles into a  
heteroheptameric pore assembly in the presence of a wild  
type (WT)  $\alpha$ HL polypeptide. Preferably, the metal-binding  
amino acid occupies a position in a transmembrane channel of  
35    the heteroheptameric pore assembly, e.g., an amino acid in

the stem domain of WT  $\alpha$ HL is substituted with a heterologous metal-binding amino acid. More preferably, the metal-binding amino acid projects into the lumen of the transmembrane channel.

5 By the term "heterologous amino acid" is meant an amino acid that differs from the amino acid at the corresponding site in the amino acid sequence of WT  $\alpha$ HL. By "analyte-binding amino acid" is meant any amino acid having a functional group which covalently or non-covalently binds to  
10 an analyte. By "transmembrane channel" is meant the portion of an  $\alpha$ HL polypeptide that creates a lumen through a lipid bilayer. The transmembrane channel of an  $\alpha$ HL pore assembly is composed of 14 anti-parallel  $\beta$  strands (the " $\beta$  barrel"), two of which are contributed by the stem domain of each  $\alpha$ HL  
15 polypeptide of the pore. By "stem domain" is meant the portion of an  $\alpha$ HL polypeptide which spans approximately amino acids 110 to 150 of SEQ ID NO:1 (see, e.g., Fig. 1F)

An  $\alpha$ HL polypeptide containing at least two non-consecutive heterologous metal-binding amino acids in a stem  
20 domain of  $\alpha$ HL is also within the invention. By "metal-binding amino acid" is meant any amino acid which covalently or noncovalently binds to a metal ion, e.g., Ser, Thr, Met, Tyr, Glu, Asp, Cys, or His. Unnatural amino acids, such as 1,2,3 triazole-3-alanine and 2-methyl histidine, which have  
25 altered  $pK_a$  values, steric properties, and arrangement of N atoms resulting in different abilities to bind metal ions, can also be introduced to confer metal-responsiveness. Preferably, the heterologous amino acids project into the lumen of the transmembrane channel, i.e., the amino acids  
30 occupy two or more of the following positions of SEQ ID NO:1: 111, 113, 115, 117, 119, 121, 123, 125, 127, 129, 131, 133, 135, 137, 139, 141, 143, 145, 147 or 149. Alternatively, the heterologous amino acids are located on

the outside of the transmembrane channel, i.e., the amino acids occupy two or more of the following positions of SEQ ID NO:1: 110, 112, 114, 116, 118, 120, 122, 124, 126, 128, 130, 132, 134, 136, 138, 140, 142, 144, 146, 148. The

5 polypeptide may contain at least three non-consecutive heterologous metal-binding amino acids in the stem domain. Preferably, the polypeptide contains at least 4 non-consecutive heterologous metal-binding amino acids in the stem domain; more preferably, the amino acids occupy

10 positions 123, 125, 133, and 135 of SEQ ID NO:1; more preferably, each these positions are occupied by the heterologous metal-binding amino acid His; and most preferably, the polypeptide is the  $\alpha$ HL mutant 4H, as described below.

15 To facilitate separation and purification of mutant analyte-responsive  $\alpha$ HL polypeptides, the polypeptide may also contain a heterologous amino acid, e.g., a Cys residue, at a site distant from the stem domain, e.g., at position 292 of SEQ ID NO:1.

20 The invention also features a heteromeric pore assembly containing a metal-responsive (M)  $\alpha$ HL polypeptide, e.g., a pore assembly which contains a wild type (WT) staphylococcal  $\alpha$ HL polypeptide and a metal-responsive  $\alpha$ HL polypeptide in which a heterologous metal-binding amino acid

25 of the metal-responsive  $\alpha$ HL polypeptide occupies a position in a transmembrane channel of the pore structure. For example, the ratio of WT and M  $\alpha$ HL polypeptides is expressed by the formula  $WT_{7-n}M_n$ , where n is 1, 2, 3, 4, 5, 6, or 7; preferably the ratio of  $\alpha$ HL polypeptides in the

30 heteroheptamer is  $WT_{7-n}4H_n$ ; most preferably, the ratio is  $WT_64H_1$ . Homomeric pores in which each subunit of the heptomer is a mutated  $\alpha$ HL polypeptide (i.e., where n = 7) are also encompassed by the invention.

Also within the invention is a digital biosensor device comprising a heteromeric  $\alpha$ HL pore assembly. The device detects binding of a metal ion to a heterologous amino acid through a single channel (single current) or  
5 through two or more channels (macroscopic current). Rather than containing a heterologous amino acid substitution, the metal-responsive  $\alpha$ HL polypeptide in the device may contain a chelating molecule associated with an amino acid in the stem domain.

10 The analyte-responsive  $\alpha$ HL polypeptides (and pore assemblies containing such polypeptides) can be used in a method of detecting the presence of an analyte, e.g., a metal such as a divalent Group IIB and transition metal. Zn(II), Co(II), Cu(II), Ni(II), or Cd(II) can be detected  
15 using the methods described herein. For example, a detection method may include the steps of (a) contacting the sample to be analyzed with an analyte-responsive  $\alpha$ HL pore assembly, and (b) detecting an electrical current in a digital mode through a single channel (single current) or  
20 two or more channels (macroscopic current). A modulation or perturbation in the current detected compared to a control current measurement, i.e., a current detected in the absence of the analyte indicates the presence (and concentration) of the analyte.

25 The invention also includes a method of identifying an unknown analyte in a mixture of analytes which includes the following steps: (a) contacting the mixture with an analyte-responsive  $\alpha$ HL pore assembly; (b) detecting an  
30 electrical current in a digital mode through a single channel (or through two or more channels) to determine a mixture current signature; and (c) comparing the mixture current signature to a standard current signature of a known analyte. A concurrence of the mixture current signature

with the standard current signature indicates the identity of the unknown analyte in the mixture.

Other features and advantages of the invention will be apparent from the following description of the preferred embodiments thereof, and from the claims. All references cited herein are incorporated by reference in their entirety.

#### Brief Description of the Drawings

Fig. 1A is a diagram showing the interpretation of a digital/stochastic response of a single channel (patch clamp) recording using an analyte-responsive  $\alpha$ HL pore assembly (the average upspike time durations  $\rightarrow K_1$ , the analyte identity; the average downspike durations  $\rightarrow [\bullet]$ , the analyte concentration).

Fig. 1B is a series of graphs of digital single channel recordings showing metal-responsiveness of an  $\alpha$ HL pore assembly at various concentrations of Zn(II).

Fig. 1C is a diagram of the structure of a heteromeric  $\alpha$ HL pore (WT<sub>7</sub>4H<sub>1</sub>) assembly showing a Zn(II) binding site with a view of the heptamer perpendicular to the seven-fold axis of the pore. The top of the structure is on the *cis* side of the membrane in bilayer experiments. The 14-strand  $\beta$  barrel at the base of the structure opens the lipid bilayer. In the 4H subunit, residues Asn123, Thr125, Gly133, and Leu135 were replaced with histidine and Thr292 with cysteine. A close-up view of the antiparallel  $\beta$  strands that contribute to the lower part of the barrel is shown in Fig. 1E below.

Fig. 1D is a diagram of the structure of a heteromeric  $\alpha$ HL pore (WT<sub>7</sub>4H<sub>1</sub>) assembly showing a Zn(II) binding site with a view of the heptamer down the seven-fold axis from the top (*cis* side) of the pore. The four heterologous histidinyl residues project into the lumen of

the channel, while Cys292 is distant from the channel mouth.

Fig. 1E is a diagram of the structure of the transmembrane channel portion of a heteromeric  $\alpha$ HL pore assembly containing the Zn(II)-responsive  $\alpha$ HL polypeptide 4H. Zn(II) is shown bound to the polypeptide at a binding site created by a heterologous metal-binding amino acid substitution.

Fig. 1F is a diagram of the structure of an  $\alpha$ HL polypeptide showing the stem domain spanning approximately amino acids 110-150.

Fig. 2A is a diagram of heteromeric combinations resulting from the assembly of mixtures of wild-type (WT) and mutant (MUT)  $\alpha$ HL monomers showing the assembly of heteromeric  $\alpha$ HL pores. The 20 different heteromers (WT<sub>n</sub>-MUT<sub>m</sub>; n=7, the total number of subunits; WT, open circles; mutant, closed circles) fall into n+1 classes categorized by the number of mutant subunits (m) in the heptamer. The proportions of heptamers (%) in each class is shown for three starting ratios of monomers (WT:MUT, 5:1; 1:1; 1:5). The values were calculated assuming that the oligomerization process does not distinguish between WT and MUT monomers, by using  $100 \cdot P_m = 100 \cdot [n! / (m! (n-m)!)] \cdot f_{MUT}^m \cdot f_{WT}^{n-m}$ , where  $f_{MUT}$  and  $f_{WT}$  are the fractions of mutant and WT subunits, respectively, in the starting monomer mix.

Fig. 2B is a diagram showing the procedure for assembly and separation of heteroheptameric  $\alpha$ HL pore assembly. Heteromers were formed from the desired ratio of WT and MUT subunits on either rabbit red blood cell membranes (rRBCM) or liposomes. The heteromers were then derivatized with IASD, which introduced two negative charges for each mutant (Cys292-containing) subunit. The eight classes of heptomer were then separated by SDS-polyacrylamide gel electrophoresis (SDS-PAGE). The members

of a particular class were obtained by elution from the polyacrylamide.

Fig. 2C is a photograph of an electrophoretic gel showing separation of different classes of  $\alpha$  heptamers.

5 WT  $\alpha$ HL and the mutant 4H, both [ $^{35}$ S]-labeled, were mixed in the ratios indicated, allowed to assemble on rRBCM and then treated as shown in Fig. 2B. The membranes were solubilized in gel loading buffer containing SDS and, without heating, subjected to electrophoresis in a 7% gel. A phosphorimager  
10 display of the molecules migrating near the 200 kDa marker (myosin heavy chain) is shown. The observed ratios of oligomer classes seen in each lane approximate those shown in Fig. 2A. The lane marked "All" contained a mixture of the solubilized samples at all five WT:4H ratios.

15 Fig. 3A is a photograph of an electrophoretic gel showing purified  $\alpha$ HL heteroheptamers. Heptamers were stable in SDS and the subunits did not interchange. All eight radiolabeled WT<sub>n-m</sub>4H<sub>m</sub> heptamers were purified by SDS-PAGE, rerun on a 40 cm long 8% SDS-polyacrylamide gel and  
20 visualized by autoradiography. The individual heteromer species (lanes 1-8) retained their relative mobilities, resulting in the staircase appearance of the image.

Fig. 3B is an electrophoretic gel showing that WT, and 4H, did not become scrambled under the conditions used  
25 for extraction, storage and reconstitution. An excised WT, band was mixed and coeluted with an excised 4H, band. The sample was kept at 4°C for 24 h and then stored at -20°C. The thawed sample was run on a 40 cm long 8% SDS  
polyacrylamide gel. The bands retained their integrity  
30 (i.e. there is no ladder of species to suggest subunit interchange).

Fig. 3C is a photograph of an electrophoretic gel showing the ratio of the WT and 4H subunits in each purified



heptamer. Heptamers were made as described in the legend to Fig. 3A. Half of each sample was subjected to electrophoresis without heating (top panel), while the other half was dissociated by heating to 95°C (bottom panel). The mutant  $\alpha$ HL monomers, modified with IASD, were separated from the more rapidly migrating WT polypeptides in a 40 cm long 10% SDS-polyacrylamide gel, allowing the quantitation of the two monomer species contained in each heptamer by phosphorimager analysis (ImageQuant, Molecular Dynamics). The expected and measured ratios are shown below each lane.

Fig. 4A is an autoradiogram of the SDS-PAGE separation of approximately 5:1 mixture of WT and 4H, from which WT<sub>6</sub>4H<sub>1</sub> was eluted and used for single channel studies. Unlabeled WT was used, so the first detectable band is WT<sub>6</sub>4H<sub>1</sub>. This band appears relatively weak here because it contains a single <sup>35</sup>S-labeled 4H subunit.

Figs. 4B, 4C, and 4D are a series of graphs of digital single-channel recordings from a heteromeric  $\alpha$ HL channel containing a Zn(II) binding site. Single-channel recordings were made using purified WT, and WT<sub>6</sub>4H<sub>1</sub> pore assemblies in planar lipid bilayers. Both cis and trans chambers of the device contained 1 M NaCl, 50mM MOPS, pH 7.5. Four consecutive traces of a single-channel current at -40mV are shown for each species. Left, currents in the presence of 100 $\mu$ M EDTA; right, currents after the addition of 150 $\mu$ M ZnSO<sub>4</sub> to the trans side of the membrane (approximately 50 $\mu$ M free Zn(II)).

Fig. 4B is a series of graphs showing digital single channel recordings using a WT<sub>6</sub> pore assembly (band 1 in Fig. 4A). The channel is open with an amplitude of -26.7pA (mean = 27.0  $\pm$  2.5 pA, n = 8). Zn(II) had no effect on the current, even when increased to 500  $\mu$ M.

Fig. 4C is a series of graphs showing digital single channel recordings using a WT<sub>6</sub>H<sub>1</sub> pore assembly (band 2 in Fig. 4A). In the presence of 100 μM EDTA the channel is open with an amplitude of -28.4 pA at -40 mV (mean =  $26.3 \pm 1.6$  pA, n = 7). The addition of 150 μM Zn(II) to the trans chamber results in discrete fluctuations between two open states, the original state (-28.4 pA) and another of -25.7 pA (mean =  $-24.4 \pm 1.8$  pA, n = 7). The ratio of the conductance of the new state to the conductance of the original state ( $g/g_o$ ) was  $0.93 \pm 0.01$  (n = 7).

Fig. 4D is a series of graphs of digital single channel recordings showing the dependence of the partial channel block of the heteromeric pore WT<sub>6</sub>H<sub>1</sub> on Zn(II) concentration. Single-channel current recordings were made at various trans free Zn(II) concentrations. A solution containing 1 M NaCl, 50mM MOPS, pH 7.5, Zn(II) was buffered with 100 μM pyridine-2,6-dicarboxylic acid and 10 μM EDTA. All points amplitude histograms are shown below each graph. The histograms can be fitted to the sum of two Gaussian functions, suggesting two distinct states: (i) the fully open channel as seen in the absence of Zn(II), (ii) the partly closed,  $g/g_o = 0.93$ , Zn(II) dependent substrate. The normalized areas of the Gaussian functions represent the occupancy of each state at the displayed Zn(II) concentration. When the openings or closing are short, the amplitudes of the transitions are underestimated, resulting in shifts of the peaks to lower values, for example, for 190nM Zn(II).

Fig. 5A is a series of graphs of digital single channel recordings from WT<sub>6</sub>H<sub>1</sub> in the presence of 5 μM free Zn(II) or 5 μM free Co(II) showing response of the heteromeric pores to different M(II)s and tuning of the sensitivity to M(II)s by adjustment of subunit composition.

Top, transmembrane potential -40 MV; bottom, transmembrane potential +40 mV.

Fig. 5B is a series of graphs showing the response of pores containing more than one 4H subunit to Zn(II).

5 WT<sub>4</sub>H<sub>2</sub> (concentration of free Zn(II) = 50  $\mu$ M), WT<sub>4</sub>H<sub>1</sub> (20  $\mu$ M) and 4H<sub>1</sub> (10 $\mu$ M). Left, digital single channel recordings of currents in the absence (EDTA) and presence of Zn(II). The zero current level is indicated (i = 0). Right, the corresponding all points histograms (light line, EDTA; dark  
10 line, Zn(II)).

Fig. 6A is a graph of a digital single channel recording from WT<sub>4</sub>H<sub>1</sub> in the presence of 150nM Zn(II).

Fig. 6B is a graph showing an expanded view of a portion of the graph in Fig. 6A.

15 Figs. 7A and 7B are graphs of digital single channel recordings from WT<sub>4</sub>H<sub>1</sub> in the presence of a solution containing 40nM Zn(II) and 40nM Ni(II) at a transmembrane potential of -40 MV.

Figs. 8A and 8B are graphs of digital single channel  
20 recordings using a pore assembly containing a 123W/135W subunit in the presence and absence of a solution containing the explosive trinitrotoluene (TNT). Fig. 8A is a recording from pores in the absence of TNT, and Fig. 8B is a recording from pores in the presence of 1 $\mu$ M TNT.

25 Description of the Preferred Embodiments  
Analyte-responsive  $\alpha$ HL polypeptides as components of biosensors

Biosensors generally have three elements: a) a  
binding site to recognize a target analyte (e.g., introduced  
30 by engineering metal-binding amino acid into an  $\alpha$ HL polypeptide to create a metal binding site in the transmembrane channel of an  $\alpha$ HL pore assembly), b) a transduction mechanism that signals the fractional occupancy

of the binding site by the analyte (e.g., salt ions flowing through the  $\alpha$ HL pore assembly/channel at a rate of 100 million/sec for the open channel compared to an altered rates when an analyte is bound), and c) a method of measurement (and processing) of the transduction signal (e.g., pA, electrical measurements of the ion flux through the  $\alpha$ HL pore assembly/channel in a membrane separating two liquid phases).

The compositions, devices and methods described herein can be used to track diverse analytes of interest in spatio-temporal gradients in water, in sediments and in the air. Such a capability would permit, for example, gradiometer-directed locomotion of robots. Other uses include detection, identification, and quantification of analytes in the environment, e.g., Cu, Zn, or Ni in effluents from underwater and dry dock hull cleaning operations, in shipboard waste processing, and in ocean micronutrient analyses.

Biosensors which incorporate protein pores as sensing components have several advantages over existing biosensors. In particular, bacterial pore-forming proteins, e.g.,  $\alpha$ HL, which are relatively robust molecules, offer all the advantages of protein-based receptor sites together with an information-rich signal obtained by single-channel recording.

$\alpha$ HL is a 293 amino acid polypeptide secreted by *Staphylococcus aureus* as a water-soluble monomer that assembles into lipid bilayers to form a heptameric pore. The heptamer is stable in sodium dodecyl sulfate (SDS) at up to 65°C. The biophysical properties of  $\alpha$ HL altered in the central glycine-rich sequence, by mutagenesis or targeted chemical modification, demonstrate that this part of the molecule penetrates the lipid bilayer and lines the lumen of

the transmembrane channel. The channel through the heptamer is a 14-strand  $\beta$  barrel with two strands per subunit contributed by the central stem domain sequence (spanning approximately amino acids 110-150 of SEQ ID NO:1).

5                    Table 1: WT  $\alpha$ HL amino acid sequence

ADSDINIKTG TTDIGSNTTV KTGDLVTYDK ENGMHKKV FY SFIDDKNHNK  
 KLLVIRTKGT IAGQYRVYSE EGANKSGLAW PSAFKVQLQL PDNEVAQISD  
 YYPRNSIDTK EYMSTLTYGF NGNVTGDDTG KIGGLIGANV SIGHTLKYVQ  
 PDFKTILESP TDKKVGWKVI FNNMVNQNWG PYDRDSWNPV YGNQLFMKTR  
 10 NGSMKAADNFL DPNKASSLL SSGFSPDFAT VITMDRKASK QQTNI DVIYE  
 RVRDDYQLHW TSTNWKGTNT KDKWTD RSSE RYKIDWEKEE MTN  
 (SEQ ID NO:1)

There is a need for biosensors that can detect a variety of analytes, ranging from simple ions to complex  
 15 compounds and even microorganisms. Protein pores made from  $\alpha$ HL polypeptides have been remodeled so that their transmembrane conductances are modulated by the association of analytes, e.g., divalent metal ions, M(II)s. The lumen of the transmembrane channel was altered to form different  
 20 analyte-binding sites by design, e.g., by using site-directed mutagenesis to insert heterologous metal-binding amino acids. An analyte-binding  $\alpha$ HL polypeptide is one that contains an engineered analyte-binding site not present in the WT  $\alpha$ HL polypeptide. An analyte-binding site can be  
 25 created by the introduction of as few as one heterologous analyte-binding amino acid, i.e., native residues may participate in forming a binding site. M(II)-binding sites can also be formed by the attachment of chelating molecules by targeted chemical modification. Combinatorial assembly  
 30 is another way to generate diversity (see Fig. 2A). M(II) detection is rapid (e.g., single channel conductance is approximately  $10^8$  ions  $\text{sec}^{-1}$ ), reversible and sensitive. With digital single-channel recording for analyte detection,

the binding sites need not be fully selective because the kinetics, extent and voltage-dependence of channel block provide a distinctive analyte signature. The voltage is gateable to further tune the biosensor. More than one  
5 analyte can be assayed simultaneously using the compositions and biosensor devices described herein. Selectivity is not a problem because a single analyte binding site can only be occupied by a single analyte at one time. Analyte-responsive  $\alpha$ HL pores have been successfully used to detect  
10 an analyte of interest, e.g., a metal ion, in a solution containing a mixture of analytes as well as in solutions containing various concentrations of a single analyte.

Digital/Stochastic Single Channel Biosensors Using Analyte-Responsive  $\alpha$ HL Polypeptides

15 The attainment of sensitivity and selectivity is a major problem with most known biosensors as they are based on an integrated signal from numerous sensor molecules. The resulting signal is analogue/steady state and contains limited information about analyte identity(ies) and  
20 concentration(s). Analogue/steady state detection data is extremely difficult to extract reliably, even by modern processing hardware and software. For example, simultaneous competition for an analyte-binding site by many different analytes is a major problem. This problem is solved by the  
25 analyte-responsive  $\alpha$ HL pores described herein.

The disclosed analyte-responsive  $\alpha$ HL compositions are unique. A biosensor using an analyte-responsive  $\alpha$ HL as the sensing component is tunable to any analyte target of  
30 interest by introducing an analyte-binding site directly into a measurable channel. Biosensors which incorporate an analyte-responsive  $\alpha$ HL pore assembly reliably detect analytes in single channel mode, i.e., an individual analyte is detected as it randomly (stochastically) hops on and off

a single binding site. These events are detected as modifications or perturbations of the ion conductance in the single channel.

A digital/stochastic biosensor device incorporating an  $\alpha$ HL pore assembly as a sensing component has several important advantages over analogue/steady state biosensors. For example, the quality of the digital signal is independent of site occupancy; therefore, the dynamic range is orders of magnitude greater. Also, rate and equilibrium constants are read directly from the averages of a few spikes providing fundamental signature information about analyte identity and concentration. Simultaneous occupancy of a single binding site by different analytes cannot occur. Instead, competing analytes appear separated in time on the signal trace, each with it's own characteristic current signature.

Fig. 1A shows the interpretation of a digital/stochastic response of a single channel (patch clamp) recording using an analyte-responsive  $\alpha$ HL pore assembly (Figs. 6A-B show an expanded view of a recording). Digital detection reports stochastic behavior of a single analyte in real chemical time. The dynamic range of a biosensor incorporating an analyte-responsive  $\alpha$ HL pore assembly is greater than 10,000 fold compared to approximately 20 fold for other known biosensors.

Structure-based design and a separation method that employs targeted chemical modification have been used to obtain a heteromeric form of the bacterial pore-forming protein  $\alpha$ -HL, in which at least one of the seven subunits contains a binding site for a divalent metal ion, M(II), which serves as a prototypic analyte. The single-channel current of the heteromer in planar bilayers is also modulated by nanomolar Zn(II). Other M(II)s (e.g., Co, Cu,

Ni, and Cd) modulate the current and produce characteristic signatures. In addition, heteromers containing more than one mutant subunit exhibit distinct responses to M(II)s. Analyte-responsive  $\alpha$ HL pores were generated through subunit  
5 diversity and combinatorial assembly.

Sensor arrays with components with overlapping analyte specificity, i.e., pore assemblies made from  $\alpha$ HL polypeptides which respond to a variety of analytes, e.g., metal ions, provide a yet more powerful means for the  
10 simultaneous determination of multiple analytes and to expand the dynamic range. By using the design principles disclosed herein, binding sites for diverse analytes, e.g., different metal ions, can be engineered into the lumen of the transmembrane channel of an heteromeric  $\alpha$ HL pore  
15 assembly or near an entrance to the transmembrane channel, e.g., near the cis entrance of the channel. The digital/stochastic detection mode can be generalized to classes of proteins other than pore-forming proteins, e.g., receptors, antibodies, and enzymes, with attached  
20 fluorescent probes to monitor individual binding events using imaging technology directly analogous to single channel recording. For example, analyte binding and dissociation from an active site (e.g., naturally-occurring or re-engineered analyte-binding site) of a remodeled  
25 fluorescent-tagged antibody, lectin, or enzyme is detected using the detection methods described above to determine the presence and/or concentration of an antigen, carbohydrate moiety, or enzyme ligand, respectively.  $\bar{z}$

The compositions and biosensor devices described  
30 herein offer sensitivity, speed, reversibility, a wide dynamic range, and selectivity in detecting and determining the identity and concentration of analytes such as metal ions.  $\alpha$ HL pores, remodeled so that their transmembrane



conductances are modulated by the association of specific analytes, make excellent components of biosensors.

Engineered pores have several advantages over existing biological components of biosensors, e.g., sensitivity is in the nanomolar range; analyte binding a rapid (diffusion limited in some cases) and reversible; strictly selective binding is not required because single-channel recordings are rich in information; and for a particular analyte, the dissociation rate constant, the extent of channel block and the voltage-dependence of these parameters are distinguishing. A single sensor element can, therefore, be used to quantitate more than one analyte at once. Furthermore, the biosensor is essentially reagentless and internally calibrated. The approach described herein can be generalized for additional analytes, e.g., small cations and anions, organic molecules, macromolecules and even entire bacteria or viruses, by introducing a binding site for any given analyte into a portion of the  $\alpha$ HL polypeptide, e.g., the stem domain, which participated in forming the transmembrane channel of the  $\alpha$ HL pore assembly. For example, a heterologous aromatic amino acid substitution can be engineered into an  $\alpha$ HL polypeptide, e.g., in the transmembrane channel portion of an  $\alpha$ HL pore assembly or at the mouth of the channel, to confer responsiveness to a variety of organic molecules. Furthermore, combinatorial pore assembly of metal-responsive  $\alpha$ HL polypeptides and WT  $\alpha$ HL polypeptides generate pores with diverse detection capabilities (see Fig. 2A).

An analyte-responsive  $\alpha$ HL pore containing a subunit in which amino acids positions 123 and 125 of SEQ ID NO:1 were substituted with tryptophan (123W/135W) was made. This mutant  $\alpha$ HL polypeptide was used to discern the presence and/or concentration of organic molecules. For example,

123W/125W binds the explosive TNT. Single-channel recordings using pore assemblies containing a 123W/125W subunit detected TNT (Figs. 8A-B).

$\alpha$ HL pore assemblies

5 WT  $\alpha$ HL pores are homomeric; that is, all seven subunits are the same. The analyte-responsive pores described herein may be homomeric or heteromeric and contain at least one mutated  $\alpha$ HL polypeptide subunit. For example, a pore assembled from seven subunits has the formula  
10  $WT_nMUT_{7-n}$ , where MUT is a mutant  $\alpha$ HL polypeptide and where  $n = 1, 2, 3, 4, 5, 6$ , or  $7$ . Preferably, the MUT subunit is an analyte-binding  $\alpha$ HL polypeptide. The amino acid sequence of MUT differs from that of WT in that MUT may be longer or shorter in length compared to the WT subunit (e.g., MUT may  
15 be truncated, contain internal deletions, contain amino acid insertions, or be elongated by the addition terminal amino acids, compared to the WT sequence); alternatively, MUT may contain one or more amino acid substitutions in the WT sequence (or MUT may differ from WT both in length and by  
20 virtue of amino acid sequence substitutions). The engineered changes in the MUT subunit preserve the ability of MUT to associate with other  $\alpha$ HL polypeptides to form a pore structure.

A heteromeric pore was made that binds the  
25 prototypic analyte Zn(II) at a single site in the lumen of the transmembrane channel, thereby modulating the single-channel current. In addition, M(II)s other than Zn(II) modulate the current and produce characteristic signatures. Heteromers containing more than one mutant subunit exhibit  
30 distinct responses to M(II)s. The invention therefore provides an extensive collection of heteromeric responsive pores suitable as components for biosensors.

### Molecular modeling of $\alpha$ HL pore assemblies

The three-dimensional structure of an  $\alpha$ HL pore assembly was determined using known methods, e.g., those described in Song et al., 1996, Science 274:1859-1865.

5 Using the modeling techniques described below, the position of amino acids which occupy the transmembrane channel portion of an  $\alpha$ HL pore assembly and/or protrude into the lumen of the transmembrane channel can be determined. For example, to analyze the structures of  $\alpha$ HL polypeptides  
10 described herein, the coordinates of carbonic anhydrase 11 (Eriksson et al., 1988, Proteins: Struct. Funct. Genet. 4:283-293) were obtained (PDB accession number 1CA3). Two  $\beta$  strands (residues 91-98 and 116-121), containing the histidines that bind Zn(II), were isolated and fitted by a  
15 blast square procedure to the  $\beta$  strands in the stem of protomer A of the  $\alpha$ HL structure (Song et al., 1996, Science 274:1859-1865). Residues 123-126 and 132-135 of  $\alpha$ HL were then replaced with 117-120 and 93-96 of carbonic anhydrase. The  $\alpha$ HL sidechains were substituted back into the structure,  
20 with the exception of the histidines at positions 123, 125, 133, and 135. The Zn(II) ion and the attached water molecule from carbonic anhydrase were left in place. In addition, Thr292 was replaced with a cysteine residue. The new molecule was drawn with Molscript (Kraulis, P.J., 1991, J. Appl. Cryst. 24:946-949) and a final version rendered  
25 with Raster3D (Merritt et al., 1994, Act Cryst. D50:869-873).

### Mutagenesis

Recombinant  $\alpha$ HL polypeptides, e.g., metal-responsive  
30  $\alpha$ HL polypeptides, were made using methods well known in the art of molecular biology. For example, the metal-responsive  $\alpha$ HL polypeptide, 4H, was made using DNA encoding a full-length  $\alpha$ HL ( $\alpha$ HL-RL) that had been partly reconstructed from

the native *S. aureus*  $\alpha$ HL gene (Walker et al., 1992, J. Biol. Chem. 267: 10902-10909) with synthetic oligonucleotides to introduce unique restriction sites in the central region (residues 116-147). Four conservative amino acid replacements are present in  $\alpha$ HL-RL: Val124 $\rightarrow$ Leu, Gly130 $\rightarrow$ Ser, Asn139 $\rightarrow$ Gln and Ile142 $\rightarrow$ Leu. The region encoding amino acids 118-138 was removed by digestion with *Bsi*WI and *Ap*al and replaced with two synthetic duplexes (*Bsi*Wi-*Sp*el and *Sp*el-*Ap*al) encoding the replacements Asn123 $\rightarrow$ His, Val124 $\rightarrow$ Leu, Thr125 $\rightarrow$ His, Gly130 $\rightarrow$ Ser, Gly133 $\rightarrow$ His, Leu135 $\rightarrow$ His. A 700 base pair fragment of the resulting construct, encompassing the four new histidines, was removed with *Nde*I and *Mfe*I and used to replace the corresponding sequence in  $\alpha$ HL-Thr292 $\rightarrow$ Cys. The entire coding region of the resulting  $\alpha$ HL-4H/Thr292 $\rightarrow$ Cys construct was verified by sequence analysis.

#### Expression and purification of $\alpha$ HL polypeptides

Monomeric WT- $\alpha$ HL was purified from the supernatants of *S. aureus* cultures using known methods, e.g., the method described in Walker et al., 1992, J. Biol. Chem. 267: 10902-10909. [ $^{35}$ S]-Methionine-labeled WT- $\alpha$ HL and  $\alpha$ HL-4H were obtained by coupled *in vitro* transcription and translation (IVTT). Separate reactions conducted with a complete amino acid premix and the premix without unlabeled methionine were mixed to yield a solution containing  $\alpha$ HL at > 10 $\mu$ g/ml.  $\alpha$ HL in the IVTT mix was partially purified by (i) treatment with 1% (w/v) polyethyleneimine (PEI) to precipitate nucleic acids, (ii) treatment with SP Sephadex C50, pH 8.0 (to remove the residual PEI), and (iii) binding to S-Sepharose Fast Flow at pH 5.2, followed by elution with 10mM sodium acetate, pH 5.2, 800mM NaCl. The concentration of  $\alpha$ HL (in the IVTT mix or after the purification) was estimated by a standard quantitative hemolytic assay.

### Oligomerization of $\alpha$ HL polypeptides

WT and  $\alpha$ HL-4H were mixed in various molar ratios (6:0, 5:1, 1:1, 1:5, and 0:6) and allowed to oligomerize on rabbit erythrocyte membranes, liposomes, and other planar bilayers. The  $\alpha$ HL polypeptides self-assemble into heteroheptameric pore assemblies in bilayers. For rabbit erythrocytes membranes, oligomerization was carried out as follows. Mixtures were incubated for 1 h at room temperature in 10mM MOPS, pH 7.4, 150mM NaCl. The membrane were washed and resuspended in 200mM TAPS, pH 9.5, treated with 0.5mM DTT for 5 min and then with 10mM 4-acetamido-4'-[(iodoacetyl)amino] stilbene-2,2'-disulfonate (IASD, Molecular Probes, Eugene, OR, USA) for 1 h at room temperature to modify the Cys292 residue on the 4H polypeptide chain. The membranes were recovered by centrifugation, taken up in gel loading buffer, without heating, and loaded onto a 7% SDS-polyacrylamide gel (40 cm long, 1.5 mm thick). Electrophoresis was carried out for 16 h at 120 V at 4°C with 0.1mM thioglycolate in the cathode buffer. The dried gel was subjected to phosphorimager or audioradiographic analysis.

### Heteroheptamer formation and purification

Heteromeric pore assembly by  $\alpha$ HL polypeptides in membranes and other planar bilayers suitable for use in biosensor devices was carried out using known methods, e.g., those described by Hanke et al., 1993, Planar Lipid Bilayers, Academic Press, London, UK; Gutfreund, H., 1995, Kinetics for the Life Sciences, Cambridge University Press, Cambridge, UK). Rugged planar bilayers are described in Cornell et al., 1997, Nature 387:580-583.

For example, to generate 4H heteroheptamers, unlabeled WT- $\alpha$ HL and  $^{35}\text{S}$ -labeled 4H were mixed in a 5:1 ratio (WT- $\alpha$ HL; 2.5 $\mu$ l of 0.5mg/ml in 20mM sodium acetate,

pH 5.2, 150mM NaCl; <sup>35</sup>S-labeled 4H; 50μl of 5μg/ml). The mixed subunits were allowed to oligomerize on liposomes for 60 min at room temperature by incubation with 10mM MOPS, pH 7.4, 150mM NaCl (26 μl) and egg yolk phosphatidylcholine (Avanti Polar Lipide, Birmingham, AL, USA; 1.5 μl of 10mg/ml). The latter had been bathed sonicated at room temperature until clear (30 min) in 10mM MOPS, pH7.4, 160mM NaCl. The mixture (60 μl) was then treated with 2 M TAPS, pH 8.5 (10 μl), and 10mM DTT (6 μl) for 10 min at room temperature, followed by 100mM IASD (5 μl in water) for 60 min at room temperature. Gel loading buffer (5x, 25 μl) was then added, without heating, and a portion (50 μl) was loaded into an 8 mm wide lane of a 40 cm long, 1.5 mm thick 6% SDS-polyacrylamide gel, which was run at 4°C at 120 V for 16h, with 0.1mM thioglycolate in the cathode buffer. The unfixed gel was vacuum dried without heating onto Whatman 3MM chromatography paper (#3030917).

Each of the eight heptamer bands was cut from the gel, using an autoradiogram as a guide. The excised pieces were rehydrated with water (100 μl). After removal of the paper, each gel strip was thoroughly crushed in the water and the protein was allowed to elute over 18 h at 4°C. The solvable eluted protein was separated from the gel by centrifugation through a 0.2 μm cellulose acetate filter (#7016-024, Rainin, Woburn, MA, USA). A portion (20 μl) was saved for single channel studies. Sample buffer (5x, 20 μl) was added to the rest of each sample. Half was analyzed, without heating, in a 40 cm long 8% SDS-polyacrylamide gel. The other half was dissociated at 95°C for 5 min for analysis of the monomer composition in a 10% gel.

#### Biosensor: planar bilayer recordings

Detection of analytes using heteroheptameric αHL pore assemblies in planar bilayers was carried out as

follows. A bilayer of 1,2-diphytanoyl-sn-glycerophosphocholine (Avanti Polar Lipids) was formed on a 100-200  $\mu\text{m}$  orifice in a 25  $\mu\text{m}$  thick teflon film (Goodfellow Corporation, Malvern, PA, USA), using standard methods, e.g., the method of Montal and Mueller (Montal et al., 1972, Proc. Natl. Acad. Sci. USA 69:3561-3566). Both chambers of the device contained 1 M NaCl, 50mM MOPS, pH 7.5, and other solutes as described in the figure legends. Two to 10  $\mu\text{l}$  of the eluted protein were added to the cis chamber to a final concentration of 0.01-0.1ng/ml. The bilayer was held at -10mV with respect to the trans side. The solution was stirred until a channel inserted. The analyte Zn(II) was added with stirring, to the trans chamber from a stock solution of 100mM ZnSO<sub>4</sub> in water. Where Zn(II) was buffered, the concentration of free Zn(II) was calculated using the program Alex (Vivadou et al, 1981, J. Membrane Biol. 122:155-175). Currents were recorded by using a patch clamp amplifier (Dagan 3900A with the 3910 Expander module), filtered at 5 kHz (four-pole internal Bessel filter) and stored with a digital audio tape recorder (DAS-75; Dagan Corporation, Minneapolis, MN, USA). For example, the data were filtered at 1-2 kHz (eight-pole Bessel filter, Model 900, Frequency Devices) and acquired at 5 kHz onto a personal computer with a Digidata 1200 D/A board (Axon Instruments). The traces were filtered at 100-200 Hz for display and analysis with the Fetchan and pSTAT programs, both of pCLAMP 6. Negative current [downward deflection] represents positive charge moving from the cis to the trans chamber.

### 30 Molecular design of heteromeric $\alpha\text{HL}$ pores

A Zn(II)-binding  $\alpha\text{HL}$  polypeptide was made by substituting one or more amino acids in the stem domain of WT  $\alpha\text{HL}$  with a heterologous metal-binding amino acid. One

example of such a Zn(II)-binding polypeptide is 4H which contains the following amino acid substitutions in the stem domain of  $\alpha$ HL: Asn123→His, Thr125→His, Gly133→His, Leu135→His, Thr292→Cys. Four histidines were introduced by mutagenesis to project into the lumen of the channel (e.g., at odd numbered positions of the stem domain) to form a cluster of imidazole sidechains.  $\alpha$ HL polypeptides in which heterologous metal-binding amino acids have been introduced such that they are located on the outside of the barrel (e.g., at even numbered positions of the stem domain) of the pore assembly also confer responsiveness to metal ions. In addition, amino acid substitutions in regions of the  $\alpha$ HL polypeptide outside the stem domain but which are close to the lumen of the transmembrane channel, e.g., at the mouth of the channel, also confer metal responsiveness.

The channel through the heptamer is a 14-strand  $\beta$  barrel with two strands per subunit (see Figs. 1C-F) contributed by the central stem domain sequence which spans approximately amino acids 110-150 of SEQ ID NO:1: EYMSTLTYGF NGNVTGDDTG KIGGLIGANV SIGHTLKYVQ (SEQ ID NO:2). Structural data indicates that the  $\beta$  barrel is sufficiently flexible for at least three sidechains to act as ligands to Zn(II) in the preferred tetrahedral configuration.

To facilitate separation of polypeptides, the 4H polypeptide was also clogged by chemical modification of the single cysteine (at position 292) with 4-acetamido-4'-[(iodoacetyl)amino]stilbene-2,2'-disulfonate (IASD). The Cys-clogged  $\alpha$ HL (Thr292→Cys; without amino acid substitutions in the stem domain) modified with IASD forms fully active homomers. This modification caused an incremental increase in the electrophoretic mobility of heptamers in SDS-polyacrylamide gels allowing heteromers to be easily separated from each other and from wild-type (WT) heptamers.



Each disulfonate made an approximately equal contribution to the mobility, which is independent of the arrangement of the subunits about the seven-fold axis. The chemical modification was distant from the stem domain of the polypeptide which lines the channel of the heteromeric pore assembly.

#### Assembly and separation of $\alpha$ HL metal-responsive heteromeric pores

There is only one possible arrangement of heteromers containing six WT and one 4H subunit ( $WT_64H_1$ ; Fig. 2A). Therefore, the  $WT_64H_1$  pore assembly is consistently and reliably formed. The 4H mutant of  $\alpha$ HL was prepared using known methods for making recombinant proteins, e.g., in vitro transcription and translation (IVTT). In some cases,  $\alpha$ HL was radiolabeled with [ $^{35}$ S]methionine. WT  $\alpha$ HL was also prepared by IVTT when labeling was desired. Alternatively, WT  $\alpha$ HL was purified from *S. aureus*. WT and 4H were mixed in a molar ratio of 5:1 and allowed to assemble on lipid bilayers, e.g., rabbit red blood cell membranes (rRBCM) or on liposomes made from egg yolk phosphatidylcholine (Fig. 2B). After assembly, the 4H subunits were modified at Cys292 with IASD. The membranes were solubilized in SDS and the heteromers separate by SDS-polyacrylamide gel electrophoresis (SDS-PAGE; Fig. 2C). Heptamers were eluted passively from the polyacrylamide with water, for reconstitution into bilayers for biophysical characterization. The eluted heteromers remained intact as shown by re-electrophoresis (Fig. 3A). These data demonstrated that the polypeptides did not become scrambled, thus indicating the predictability of pore assembly. For example,  $WT_7$ , and  $WT_64H_2$  were not formed from  $WT_64H_1$ .

In two out of five such runs, small amounts of monomer (<5%) were detected. Such breakdown was probably

due to the storage conditions that the two samples experienced (e.g. for the sample displayed in Fig. 3A, several freeze/thaw cycles, followed by storage at 4°C for ten days). In the other three runs, where freshly eluted heptamers were examined, monomers were not detected at all. In a definitive experiment, gel slices containing the homomers WT, and IASD-modified 4H, were mixed and taken through the elution and storage procedures before re-electrophoresis, which again indicated no scrambling (Fig. 3B). Furthermore, the eluted heptamers were free of residual proteins from the IVTT mix, as determined by silver staining. Finally the ratio of the  $\alpha$ HL polypeptides in each of the heteromeric pore assemblies examined was as expected, when determined by quantitative analysis of radio-labeled polypeptides from purified heteromers dissociated by heating to 95°C (Fig. 3C). The electrophoretic gel shown in Fig. 4A confirms the heteromeric channel structure of the  $\alpha$ HL pore assembly.

Digital single-channel currents from heteromeric metal-responsive pores

The properties of WT<sub>4</sub>H<sub>1</sub> were examined by digital single-channel recording in a planar bilayer biosensing apparatus. Methods for forming planar bilayers in biosensors are known in the art, e.g., Hanke et al., 1993, Planar Lipid Bilayers, Academic Press, London, UK or Gutfreund, H., 1995, Kinetics for the Life Sciences, Cambridge University Press, Cambridge, UK. In this experiment, a lipid bilayer was formed across an aperture (100-200  $\mu$ m diameter) in a teflon film (25  $\mu$ m thick) that separates two chambers (2ml each) containing electrolyte. With a potential applied across the bilayer, the ion flux through single  $\alpha$ HL pores was measured with a sensitive, low-noise amplifier.

To obtain single-channel currents, the eluted heptamers were added at high dilution (typically 1:1000) to the *cis* chamber of the bilayer apparatus to a final concentration of 0.02-0.1ng/ml (Figs. 4B-D). WT<sub>6</sub>4H<sub>1</sub> exhibited a partial and reversible channel block ( $g/g_0 = 0.93 \pm 0.01$ ;  $n=7$ ) in the presence of 50  $\mu$ M Zn(II) in the trans compartment with the transmembrane potential held at -40mV (Fig. 4C).

The behavior of heteromeric WT<sub>6</sub>4H<sub>1</sub> pores were compared to two different control pores. WT<sub>7</sub> control pores were not sensitive under the conditions described above (see Fig. 4B) and were unaffected by up to 500  $\mu$ M Zn(II). Heteromeric WT<sub>6</sub>4H<sub>1</sub> pores were also compared to control pores made with WT<sub>6</sub>4H<sub>1</sub> with an additional Thr292→Cyr mutation, modified with IASD. This heptamer also gave no response with Zn(II). Fig. 4B shows data from WT<sub>7</sub>,  $\alpha$ HL pore assemblies (i.e., control pores); control pores did not respond to the presence of Zn(II) or EDTA (a chelating agent that complexes M(II)).

Analysis of conductance histograms for WT<sub>6</sub>4H<sub>1</sub> obtained for a series of buffered Zn(II) concentrations (Fig. 4D) yielded an EC<sub>50</sub> for trans Zn(II) of  $112 \pm 23$  nM ( $n=3$ ). The EC<sub>50</sub> is the concentration of free M(II) that effects 50% occupancy of the binding site of 4H. Kinetic analysis of the current traces yielded a second-order associated rate constant ( $k_{on}$ ) for Zn(II) of  $3.2 \pm 0.4 \times 10^8 \text{ M}^{-1} \text{ s}^{-1}$  ( $n=4$ ) which approaches the diffusion limit, and a dissociation rate constant ( $k_{off}$ ) of  $33 \pm 2 \text{ s}^{-1}$  ( $n=4$ ). The EC<sub>50</sub> value was lower than expected for two histidinyl ligands and approached the values found for structures with three histidines with favorable geometry (e.g. 36nM for a mutated retinol-binding protein), suggesting that a modest distortion of the  $\beta$  barrel can be tolerated that places at

least three of the four histidines in conformations suitable for coordination of the bound metal. The flexibility of the barrel is supported by (1) the three-dimensional structure of  $\alpha$ HL, (2) the fact that for  $\alpha$ HL in liposomes blue shifts of the fluorescent probe acrylodan (attached at single cysteine residues in the  $\beta$  barrel) do not alternate with residue number (as would be required for nondistorted  $\beta$  strands), and (3) the existence of mutants with proline residues in the central domain that form pores.

The conductance of WT<sub>7</sub> pores ( $675 \pm 62$  pS, 1M NaCl, 50mM MOPS, pH7.5, -40mV,  $n=8$ ) was similar to that of WT<sub>6</sub>4H<sub>1</sub> in the absence of Zn(II) ( $660 \pm 40$  pS,  $n=7$ ). The conductance of WT<sub>6</sub>4H<sub>1</sub> with Zn(II) bound was reduced to  $610 \pm 45$  pS ( $n=7$ ). A partial channel block may be due to a simple physical blockade, distortion of the barrel, or electrostatic effects.

Figs. 4C and 4D show digital responses of the engineered WT<sub>6</sub>4H<sub>1</sub> hybrid channel to various levels of Zn(II). The digital pattern is due to the stochastic (random) effect of single zinc ions hopping on and off the tetra-histidyl binding site engineered into the lumen of the transmembrane channel of an  $\alpha$ HL pore assembly. The two channel states are open (Zn(II) off, 100% open) and gated (Zn(II) on, 93% open). Average time in the open state is the reciprocal of bimolecular rate constant  $\times$  [Zn(II)], from which Zn(II) is quantified, while average time in the gated state is the reciprocal of the first order off constant (the analyte signature or identity). Monovalent metal cations gave no signal. These data indicate that the metal-responsive  $\alpha$ HL polypeptides and pore assemblies used as components of a biosensor provide a means to achieve unambiguous analyte identity and concentration(s). Existing chemo/bio-sensors are analog/steady state, whereas the channel of the  $\alpha$ HL pore assembly is digital/stochastic.

Fig. 4 also shows that  $\alpha$ HL pore assemblies have an wide dynamic range of analyte detection (at least 10,000-fold in analyte concentration. Even at very low fractional site occupancies, the signal (being digital and not analog) is not degraded. At very low site occupancy, it simply may take longer to collect to collect data (however, sensitivity and selectivity is not compromised).

Metal-responsive  $\alpha$ HL pores produce characteristic single-channel signatures in response to various divalent metal cations

To determine whether  $WT_4H_1$  can distinguish between different  $M(II)$ s, the effects of  $Co(II)$ ,  $Ni(II)$  and  $Cu(II)$  on single-channel currents were examined. Each gave a characteristic current signature. For example, at -40mV 5 $\mu$ M,  $Co(II)$  produced a distinctive current signature compared to, e.g.,  $Zn(II)$  (Fig. 5A, top). At higher  $Co(II)$  concentrations, the signal was continuous resulting from the rapid interconversion of three states, one with higher conductance than  $WT_4H_1$  in the absence of  $M(II)$ . At +40mV, two states were seen with 5 $\mu$ M  $Co(II)$  (Fig. 5A, bottom). The effect on current amplitude is similar to that of  $Zn(II)$  at this membrane potential, but the rates of  $Co(II)$  association and dissociation are considerably slower. These data also show that the responses of single-channel currents to membrane potential contain additional information about the concentration and identity of analytes.

The data in Fig. 5A indicate that different  $M(II)$  give different digital output patterns, i.e., spikes from one  $M(II)$ , e.g.,  $Zn(II)$ , are not hidden under the spikes of another, e.g.,  $Co(II)$ , because only one metal ion can occupy a single site at one time. In a complex mixture of analytes, deconstruction of the signal is required to isolate the current signature of an analyte of interest.

The sensitivity and precision of analyte identification achieved by the compositions and digital/stochastic devices of the invention vastly exceed those achieved by known analogue/steady state biosensor devices. For example, simultaneous competitive inhibition owing to incomplete selectivity is a universal problem with conventional chemo/bio-sensors, requiring extensive down-stream processing. In contrast, the identity and concentration of analytes can easily, reliably, and accurately determined from traces such as those in Figs. 4A-D and 5A-B, i.e., analytes can be identified (as well as quantified) by the single-channel current signature ( $\Delta g$ ,  $k_{on}$ ,  $k_{off}$ , voltage dependence of these parameters). Fig. 5 also illustrates that the channel can further be tuned by changing the transmembrane voltage. Figs. 7A-B show that digital output patterns corresponding to different analytes allow the detection and quantification of analytes, e.g., Zn (II) and Ni (II), even in solutions containing a mixture of analytes. These data indicate that  $\alpha$ HL biosensors may be used to detect, identify, and quantify analytes in complex mixtures, e.g., environmental samples or waste water samples.

Additional 4H heteromers exhibit different responses to divalent cations

Structural variants of  $\alpha$ HL pores resulting from combinatorial assembly provide yet another means by which to tune an  $\alpha$ HL channel for detection of analytes. In addition to the experiments described above, other combinations of  $WT_n4H_m$  were tested. The extent of single-channel block by Zn(II) increased with the number of 4H subunits. Multiple subconductance states were observed as exemplified by the data for  $WT_54H_2$ ,  $WT_44H_3$ , and  $4H_7$  (Fig. 5B). The specific permutations of the  $WT_54H_2$  and  $WT_44H_3$  pores in these recordings was not determined, however single-channel